BRIEF COMMUNICATION

Giant Magnetoresistance in Self-Doped $La_{1-x}MnO_{3-\delta}$ Thin Films¹

S. Sundar Manoharan, Dhananjay Kumar, and M. S. Hegde

Solid State and Structural Chemistry Unit, Indian Institute of Science, Bangalore 560 012, India

K. M. Satyalakshmi

Department of Metallurgy, Indian Institute of Science, Bangalore 560 012, India

and

V. Prasad and S. V. Subramanyam

Department of Physics, Indian Institute of Science, Bangalore 560 012, India

Received February 15, 1995; accepted March 3, 1995

Highly textured La_{0.7}MnO_{3- δ} films were grown on LaAlO₃ (100) substrates for the first time. The films showed a metal-insulator transition at 220 K and exhibited giant magnetoresistance with $\Delta R/R_0 = 85\%$ and $\Delta R/R_H > 550\%$. This study demonstrates that the La_{1-x}MnO_{3- δ} films are unique among the other La_{1-x}MnO₃ (M = Ca, Ba, and Pb) thin films showing giant magnetoresistance by virtue of being a self-doped system. © 1995 Academic Press, Inc.

Giant magnetoresistance effects (GMR) in multilayer superlattice thin films containing ferromagnetic layers separated by nonmagnetic layers have generated considerable interest in magnetic layered materials due to their potential use as the sensor element in a magnetoresistive "read" head for information storage systems (1-3). Although magnetoresistance (MR) was observed earlier in Ti_2O_3 (4) and $La_{1-x}Pb_xMnO_3$ (5) single crystals, Baibich et al. (1) were the first to report GMR in (Fe/Cr)n multilayer thin films. Subsequently, GMR in oxide systems such as Nd_{0.5}Pb_{0.5}MnO₃ (NPMO) (6) single crystals, in thin films of $La_{0.66}$ $Ba_{0.34}MnO_3$ (LBMO) (7), $La_{0.72}$ $Ca_{0.28}MnO_3$ (LCMO) (8), $La_{0.6}Pb_{0.4}MnO_3$ (LPMO) (9), $La_{1-x}Sr_xMnO_3$ (10), and in bulk $La_{1-x}Sr(Ca)_xMnO_3$ systems (11) were reported. A recent report by Jin et al. (12) generated further interest in LCMO films which show a decrease of over 97% in $\Delta R/R_0$. The reason for such colassal GMR is attributed to the ferromagnetic ordering of Mn-O₂ layers in the crystallographic a-b planes, separated by nonmagnetic La-(Ca)-O layers, and to antiferromagnetic ordering along the c-axis. The presence of mixed valence in Mn (Mn³⁺ and Mn⁴⁺) is responsible for the occurrence of both ferromagnetism and metallic conductivity.

To field and Scott (13) have compared the oxygen nonstoichiometry of the parent LaMnO_{3+ δ} with that of other perovskite oxides, LaMO₃, where M = V, Cr, Fe, and Co. Their study indicated that LaMnO_{3+ δ} is apparently the only system in the lanthanum transition metal perovskites which exhibits a wide range of oxidative nonstoichiometry. The neutron diffraction study of LaMnO_{3+ δ} indicated that any increase in oxygen content over 3.0 results in the oxidation of LaMnO₃ to a defect perovskite and in a partial elimination of La₂O₃, leaving vacancies principally on the lanthanum sites, as per the schematic equation:

$$LaMnO_{3,12} \rightarrow 0.02La_2O_3 + La_{0.96}MnO_3$$
. [1]

Although studies on the bulk LaMnO₃₊₈ (14) show magnetic ordering at temperatures as high as 250 K, it is not clear in any of these studies how the extra oxygen is accommodated in the perovskite-related structure. However, LaMnO₃₊₈ decomposing into La_{1-x}MnO₃ and La₂O₃ as in Eq. [1] leads to partial oxidation of Mn³⁺ and Mn⁴⁺, which is equivalent to La_{1-x}M_xMnO₃ (M, divalent ion) oxides so far as the idea of doping is concerned. It is noteworthy that Goodenough (15) has reported ferromagnetic ordering and an M-I transition in both La_{1-x}MnO₃ and La_{1-x}M_xMnO₃, where the Curie temperature, T_c , varies from 166 K for La_{1-x}MnO₃ to 270 K < T_c < 375 K in the La_{1-x}M_xMnO₃ systems (M, divalent ion).

¹ Contribution 1091 from the Solid State and Structural Chemistry Unit.

It was in this context that we thought it is interesting to study the $\text{La}_{1-x}\text{MnO}_3$ system in bulk as well as thin film forms. In this report we document a metal-insulator transition in the self-doped $\text{La}_{1-x}\text{MnO}_{3-\delta}$ system in the bulk solid. We also report, for the first time; thin films of self-doped $\text{La}_{1-x}\text{MnO}_{3-\delta}$ showing a GMR of 85% for $\Delta R/R_0$ and of over 550% for $\Delta R/R_H$ where R_0 and R_H are the resistances in zero-field and in an applied field.

The bulk sample of nominal composition La_{0.7}MnO₃ was prepared by the ceramic method. The component oxides La₂O₃ and MnCO₃ were mixed in stoichiometric amounts and ground in alcohol. The powder was dried, reground, and heated at 1050°C for 24 hr. Shown in Fig. 1 (curve a) is the X-ray diffraction pattern of the bulk solid recorded using a JEOL-JDX 8p powder diffractometer with Cuκα radiation. The compound crystallizes in the rhombohedral structure, which can be indexed for a hexagonal unit cell with a = 5.514(4) Å and c = 13.319(2)A. The lattice parameters agree well with recent powder X-ray and neutron diffraction studies (16) for $La_{1-x}MnO_{3-\delta}(x=0.15)$. All the lines in the X-ray pattern could be indexed to the $La_{1-x}MnO_{3-\delta}$ phase, except for a weak line (at $2\theta = 36.2^{\circ}$) corresponding to Mn₃O₄, which, from relative intensities, was estimated to be about 4% in abundance. The total oxygen estimation was carried out by iodometric titration. By taking into account the amount of Mn₃O₄ present in the sample the chemical analysis leads to the composition La_{0.76}MnO_{2.73}. The corresponding charge distribution of the cations in the oxide

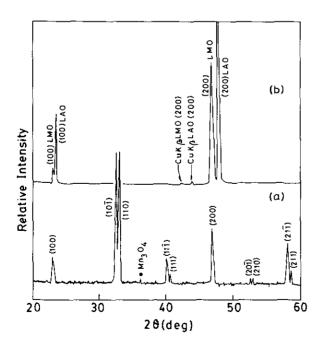


FIG. 1. X-ray diffraction pattern of (a) a polycrystalline sample of $La_{0.76}M\pi O_{2.73}$ (LMO) without Ag addition and (b) an as-deposited LMO thin film on LaAlO₃ substrate without the addition of Ag. XRD pattern of LMO film with Ag addition was identical to curve b.

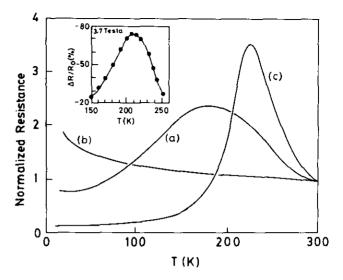


FIG. 2. R versus T plot of the (a) LMO polycrystalline sample without the addition of Ag (b, c) LMO thin film without and with the addition of Ag, respectively. Inset shows the plot of MR ratio versus temperature at 3.7 T.

can be represented by $La_{0.76}^{3+} Mn_{0.82}^{3+} Mn_{0.18}^{4+} O_{2.73}$, which clearly shows 18% of Mn to be in the tetravalent state.

The resistance versus temperature (R vs T) curve of the bulk sample is shown in Fig. 2 (curve a). The room temperature resistivity, ρ , of the bulk sample was 6 m Ω cm. The compound exhibits a clear maximum at 180 K. Magnetic susceptibility measurements confirmed the ferromagnetic ordering at the peak, which is often considered to be indicative of an M-I transition. Such behavior is in line with alkaline-earth-metal-doped LaMnO₃ systems (15). This implies that partial conversion of Mn³⁺ to Mn⁴⁺ is caused by vacancies at the lanthanum and oxygen sites; thus, this compound represents a self-doped system.

The observation of an M-1 transition with ferromagnetic ordering in bulk, La_{0.76}MnO_{2.73} (LMO), prompted our efforts to fabricate thin films by pulsed laser deposition. A bulk LMO sample was used as the target for the deposition of LMO films on LaAlO₃ (100) substrates by the laser deposition technique, employing a KrF excimer laser ($\lambda = 248$ nm, reprate = 5 Hz) with a pulse energy density of 1.8 J/cm² (of 350 mJ). The substrate temperature was maintained at 720°C, with an O₂ pressure of about 550 mTorr during the deposition. The films were cooled under 300 Torr of oxygen after the deposition.

The LMO films were ≈ 2500 Å in thickness. The X-ray diffraction study of the film shows a highly textured growth of the LMO film on the LaAlO₃ substrates, as shown in Fig. 1 (curve b). Reflections corresponding to (100) and (200) lines of LMO were observed, along with corresponding LaAlO₃ peaks, indicating that the growth is cubic with the lattice parameter a = 3.86 Å, close to the pseudo cubic "a" parameter of bulk LMO.

Although the R vs T plot for the bulk LMO target showed a peak at 180 K (Fig. 2, curve a), the as-deposited film of this composition exhibited weak semiconducting behavior (Fig. 2, curve b) down to 15 K. This discrepancy may be associated with the inadequate oxidation of Mn³⁺ to Mn⁴⁺ ions in the films, due to oxygen deficiency during deposition. The oxygen deficiency in the films, in turn, is caused by the dominance of O₂-outdiffusion over O₂indiffusion during growth of these films at elevated temperature (720°C). To promote O₂-indiffusion, 5% Ag₂O was added to the bulk LMO target and sintered at 800° for 24 hr. This target was used for the thin film deposition. The powder XRD pattern of the Ag added LMO target was found to be similar to bulk LMO without Ag. Transmission electron microscopy showed the presence of Ag metal in La_{1-r}MnO₃. During the ablation of the Ag-metalcontaining target, the silver becomes oxidized to AgO; during film growth the AgO dissociates, yielding its oxygen to the LMO film. It is noteworthy that addition of Ag₂O to bulk YBa₂Cu₃O_{7-x} (YBCO) increases the oxygen content in the YBCO film (17).

It was gratifying that the LMO film containing Ag indeed showed the same metal-insulator behavior as that of the bulk sample (curve c in Fig. 2). The inset of Fig. 2 shows the variation of MR ratio with temperature at a magnetic field of 3.7 T. The maximum at 210 K indicates ferromagnetic ordering which lies to the left of the M-I transition temperature, as observed by Jin et al. (12) for the LCMO thin films.

Magnetoresistance measurements were carried out using a superconducting magnet with a maximum applied

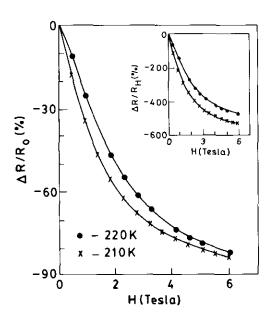


FIG. 3. Plot of magnetoresistance ratio versus magnetic field at different temperatures. Inset shows a plot of $\Delta R/R_H$ versus magnetic field.

field of 6 T, in both the parallel and perpendicular directions to the film surface. The GMR values of the LMO film measured in both directions to the applied magnetic field were of the same magnitude, indicating that the system is nearly isotropic. Shown in Fig. 3 are the MR curves with the magnetic field perpendicular to the film. The most striking feature is the magnitude of the $\Delta R/R_0$ ratio, which is 85% of GMR at 210 K, while $\Delta R/R_H = 550\%$, as shown in the inset of Fig. 3. This value for the as-deposited LMO films is quite significant as compared with the recent report on LCMO systems, where $\Delta R/R_0$ is reported to be 97% at 77 K. From our studies, it is clear that the phase giving rise to magnetic ordering both in bulk and in film is $La_{1-x}MnO_{3-\delta}$, with an estimated lanthanum and oxygen defect concentration of x = 0.24 and $\delta = 0.27$, respectively. This suggests that it is the vacancies in lanthanum and oxygen sites that cause such a large change in resistance.

In summary, we have presented the first observation of an M-I transition and ferromagnetic ordering in the bulk, and the fabrication of self-doped LMO thin films on LaAlO₃ substrate showing GMR = 85% in the as-deposited films. The observation of such large magnitude of GMR in the self-doped LMO films is attributed to lanthanum and oxygen vacancies.

ACKNOWLEDGMENTS

The authors thank Professor J. Gopalakrishnan for useful discussions. Thanks are due to Dr. N. Y. Vasanthacharya for his help in the magnetic susceptibility measurements. One of the authors (K.M.S) thanks CSIR, New Delhi, for the award of a fellowship. Financial assistance from the Department of Science and Technology, Government of India, is gratefully acknowledged.

REFERENCES

- M. N. Baibich, J. M. Broto, A. Fert, F. Nguyen Van Dau, F. Petroff P. Etienne, G. Creuzet, A. Frederick, and J. Chazelas, *Phys. Rev. Lett.* 61, 2472 (1988).
- 2. P. M. Levy, Science 256, 972 (1992).
- 3. R. L. White, IEEE Trans. Magn. 28, 2482 (1992).
- L. L. Van Zandt, J. M. Honig, and J. B. Goodenough, J. Appl. Phys. 39, 594 (1968).
- 5. C. W. Searle and S. T. Wang, Can. J. Phys. 48, 2024 (1970).
- R. M. Kusters, J. Singelton, D. A. Keen, R. McGreevy, and W. Hayes, *Physica B*. 155, 362 (1989).
- R. Von Helmolt, J. Wecker, B. Holzappel, L. Schultz, and K. Samwar, Phys. Rev. Lett. 71, 2331 (1993).
- K. Chahara, T. Ohno, M. Kasai, and Y. Kozono, Appl. Phys. Lett. 63, 1990 (1993).
- 9. S. Sundar Manoharan, et al., J. Appl. Phys. 76, 3923 (1994).
- H. L. Ju, C. Kwon, Qi Li, R. L. Greene, and T. Venkatesan, Appl. Phys. Lett. 65, 2108 (1994).
- R. Mahesh, R. Mahendiran, A. K. Raychaudhuri, and C. N. R. Rao, J. Solid State Chem. 114, 297 (1995).
- S. Jin, T. H. Tiefel, M. McCormack, R. A. Fastnacht, R. Ramesh, and L. H. Chen, Science 264, 413 (1994).
- 13. B. C. Tofield and W. R. Scott, J. Solid State Chem. 10, 183 (1974).

BRIEF COMMUNICATION

Lett. 62, 3522 (1993).

- 14. J. A. M. Van Roosmalen, E. H. P. Cordfunke, R. B. Helmholdt and H. W. Zandberger, J. Solid State Chem. 110, 100 (1994); J. A. M. Van Roosmalen and E. H. P. Cordfunke, J. Solid State
- Chem. 110, 109 (1994).

15. J. B. Goodenough, Prog. Solid State Chem. 5, 145 (1971).

- - 16. S. Habekost, P. Norby, J. E. Jorgensen, and B. Lebech, Acta. Chem. Scand. 48, 377 (1994).
 - 17. Dhananjay Kumar, M. Sharon, R. Pinto, P. R. Apte, S. P. Pai, S. C. Purandare, L. C. Gupta, and R. Vijayaraghayan, Appl. Phys.